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STRUCTURES OF ARYLSELENENYL IODIDES: THE ROLE OF ARYL STACKING AND "SOFT-SOFT" SE" I AND I" I INTERACTIONS

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The influence of packing effects and Se...I / I...I interactions on the stability of Iodoselanes are discussed.

Keywords: selenium, iodoselanes, secondary interactions, packing effects

1a, 1b R_2 , R_4 , $R_6 = t$ - Bu // R_3 , $R_5 = H$ **2a, 2b** R_2 , R_4 , $R_6 = i$ - Prop // R_3 , $R_5 = H$

3a, 3b R_2 , R_4 , $R_6 = Me // R_3$, $R_5 = H$

4a, **4b** R_2 , R_3 , R_5 , $R_6 = Me // R_4 = H$

When 2,4,6-t-Bu₃C₆H₂SeI **1a** was characterised as first stable solid selenenyl iodide, it was assumed, that its existence was due to steric destabilisation of the congested diselenide (2,4,6-t-Bu₃C₆H₂Se), **1b** [1].

The less crowded selenenyl iodide 2,4,6-i-Pr₃C₆H₂SeI 2a, exists in solution in equilibrium with its parent diselenide (2,4,6-i-Pr₃C₆H₂Se)₂ 2b. However, 2a cannot be isolated. The solid product from 2a/2b is (Fig.1) an inclusion compound of two equivalents of 2b with one I₂ molecule ^[2].

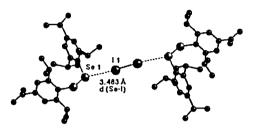


FIGURE 1: Iodine inclusion between two molecules of (2,4,6-i-Pr₃C₆H₂Se)₂ (2b)

Like 1a and 2a, even less crowded 2,4,6-Me₃C₆H₂SeI 3a exists in solution in equilibrium with its parent diselenide $(2,4,6-Me_3C_6H_2Se)_2$ 3b. In this case, however, the selenenyl iodide 3a was isolated as pure solid.

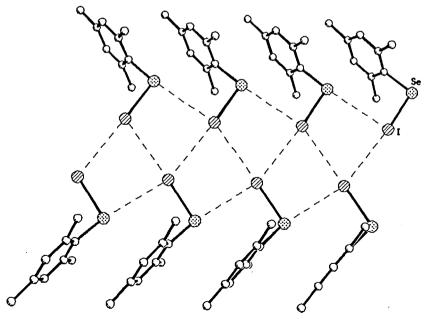
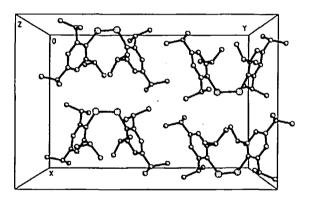


FIGURE 2: Interactions within the layers of 2,4,6-Me₃C₆H₂SeI (3a); [d(Se-I): 383.9 pm; d(I-I): 384.1 pm]

In the solid state structure of selenenyl iodide 3a (Fig.2) aryl stacking of the mesityl groups is accompanied by significant intermolcular Se⁻⁻I and I⁻⁻I contacts (L-type patterns related to those of $(I_2)_2$).

The comparison of the structures of diselenide 2b and 3b is very interesting (Fig.3): The 2,4,6-i-Pr₃C₆H₂Se substituents of 2b do not allow fair aryl stacking but allow I₂ intercalation! The structure of diselenide 3b, however, involves slipped-sandwich-like aryl stacking of the mesityl groups.

2b



3 b

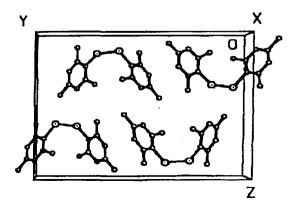


FIGURE 3: Packing of Tip₂Se₂ (2b) in comparison with Mes₂Se₂ (3b)

The structure of solid 2,3,5,6-Me₄C₆HSeI 4a is related to that of 3a (Fig.4): the stability of the Se-I bond in the solid compound 4a (in equilibrium with the diselenide (2,3,5,6-Me₄C₆HSe)₂ 4b in toluene solution) will to a significant part be due to favourable intermolecular soft-soft interactions within the columnar structure. Channels of weak soft-soft interactions between Se-I moities of the molecules are bridged by staples of aromatic durene groups.

It appears, that not only steric destabilisation of molecular diselenides may favour the existence of molecular R-Se-I compounds, but also "packing effects" i.e.: cooperative Se^{-I}I and I^{-I}I interactions combined with favourable aryl stacking.

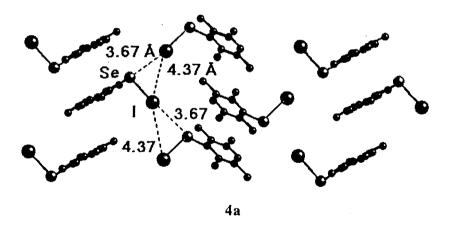


FIGURE 4: Columnar packing of 2,3,5,6-Me₄C₆HSeI (4a) with closest intermolecular Se...I and I...I distances; Bondlength Se-I: 2.53 Å

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[2.] W.-W. du Mont, A. Martens-von Salzen, Phosphorus, Sulfur, and Silicon, Vol. 67, 1-4 (1992), 67 - 71